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A BIBLIOGRAPHY OF THERMOPHYSICAL PROPERTIES OF ARGON FROM 0 TO 300°K

L. A. HALL, J. G. HUST, AND A. L. GOSMAN



U. S. DEPARTMENT OF COMMERCE
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A BIBLIOGRAPHY OF THERMOPHYSICAL PROPERTIES OF ARGON
FROM 0 TO 300°K¹

L. A. Hall, J. G. Hust, and A. L. Gosman

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Author

A bibliography of 450 references is presented for mechanical, thermodynamic, and transport properties of argon from 0 to 300°K. Each article has been reviewed and coded with regard to properties studied, type of article (i.e., experimental, theoretical, etc.), and method of presentation of the data. The temperature and pressure ranges for each property under consideration are also given. An index was prepared according to property with 4 sub-categories: solid, liquid, gas up to 200°K, and gas above 200°K. ABST

1. INTRODUCTION

The Compilation Unit of the Cryogenic Data Center has in its mission the critical evaluation of quantitative information from the world's literature related to the thermophysical properties of materials at cryogenic temperatures. At the outset of the study of a particular material, copies of the documents concerned with the properties are obtained and reviewed. As the task of document accumulation continues, a concerted effort is made to complete a systematic and thorough literature search on the selected topic. For argon, the collection of documents began over five years ago in conjunction with the data compilation presented in the "Compendium".² This bibliography on the properties of argon is the publication of the results, which provides a methodical organization of the literature on argon.

The initial literature search was conducted by the use of various abstracting journals, and in particular of Chemical Abstracts. Copies of each of the pertinent documents were obtained, and each document was reviewed and coded. In addition, all pertinent articles, which were referenced in these documents, were also obtained, reviewed, and coded. The search of Chemical Abstracts has been periodically up-dated.

¹ This bibliography is a result of a study made under contract with the National Aeronautics and Space Administration.

² A Compendium of the Properties of Materials at Low Temperature (Phase I.) Part I. Properties of Fluids, V. J. Johnson, editor, Wright Air Development Division Tech. Rept. 60-56 (1960) 560 p.

2. FORMAT FOR LISTING CITATION AND DOCUMENT CONTENTS

The citations have been arranged alphabetically by first author and numbered. Late additions were inserted in alphabetical sequence and given the number of the preceding article followed by an A, B, etc. Only information from the article which concerns the properties of argon was noted in this bibliography.

The information given for each citation includes and is ordered as follows:

1. Author(s)
2. Title (original language) and translated title, if original is in a language other than English
3. Reference (If the same article is published in more than one place, each reference is cited.)
4. Chemical Abstract number or ASTIA number, when they are known
5. Properties studied for argon, state of substance, temperature and pressure ranges as available
6. Substances, other than argon, which are discussed in the article (Where no fluids are listed, the contents of the document are restricted to properties of argon only.)
7. Designation as to primary character of article
 - a. experimental
 - b. theoretical
 - c. compilation¹
 - d. correlation
 - e. reference book²
8. Form in which data is reported
 - a. tabular - tables (number of values)
 - b. graphical
 - c. equations
9. Apparatus, if described or illustrated
10. Original language, if other than English.

¹ In compilation, the bibliography number of the original article from which the data was obtained is listed.

² The amount of data in reference books is not given.

3. INDEX OF PROPERTIES

The bibliography is indexed according to property with sub-indexes for the state of the substance; i.e., solid, liquid, gas up to 200°K, and gas above 200°K. The letters E, T, C, and R following each citation number refer to the type of data; i.e., E = experimental, T = theoretical, C = compilation, correlation, calculation, and R = review, discussion, reference work. A few reference books were coded by property only and are designated by *R following the citation number.

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† The properties for which the literature search has been conducted, including a complete search in Chemical Abstracts (1917 to 1962).

PROPERTY INDEX (CONT.)

2. EQUATION OF STATE, VIRIAL COEFFICIENTS

2 T	22 T	23 T	24 T	25 T	26 T	27 T	31 C	32 T	34 T
55 T	60 E	67 T	80 T	88 T	89A T	92 T	105 R	110 C	147 T
160 T	164 T	165 T	167*R	171 T	172 T	178 C	195 T	197 T	198 T
199 T	204 C	205 E	216 T	231 T	234 T	236 E	250 R	254 †	265 E
274 E	278 R	289 C	290 E	292 T	328 C	334 R	364 T	364A T	365 T
381 E	385 E	394 E	395 T	399 T	402 E	423 T	425 T		

SOLID

32 T	92 T	105 R	160 T	165 T	198 T	199 T	423 T
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LIQUID

2 T	231 T	236 E	385 E	399 T
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GAS (UP TO 200 DEGREES K)

24 T	25 T	31 C	60 E	67 T	88 T	164 T	178 C	204 C	205 E
234 T	236 E	254 T	289 C	290 E	292 T	334 R	364 T	364A T	365 T
381 E	395 T	399 T							

GAS (ABOVE 200 DEGREES K)

24 T	25 T	26 T	31 C	34 T	60 E	67 T	88 T	164 T	171 T
172 T	205 E	236 E	250 R	254 T	274 E	289 C	290 E	292 T	334 R
364 T	364A T	365 T	381 E	395 T	399 T	402 E			

PROPERTY INDEX (CONT.).

3. EXPANSIVITY AND COMPRESSIBILITY

15 E	16 C	28 T	32 T	33 T	89A T	100 E	103 E	105 R	106 T
107 T	147 T	160 T	189*R	218 T	219 T	229 E	232 E	239 T	241 E
265 E	285 E	317B T	322 E	329 E	351 E	356 E	357 E	376 R	377 E
383 E	385 E	386 E	396*R	399 T	413 T	422 T	425 T	426 T	

SOLID

15 E	16 C	28 T	32 T	33 T	89A T	100 E	103 E	105 R	106 T
160 T	229 E	351 E	356 E	357 E	413 T	422 T			

LIQUID

15 E	229 E	241 E	376 R	377 E	383 E	385 E	386 E	399 T	413 T
------	-------	-------	-------	-------	-------	-------	-------	-------	-------

GAS (UP TO 200 DEGREES K)

265 E	322 E	399 T	413 T
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GAS (ABOVE 200 DEGREES K)

232 E	265 E	285 E	322 E	329 E	399 T	413 T
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4. VAPOR PRESSURE

SOLID

41 E	45 E	70 E	85 E	86 E	98*R	99 R	115 R	127 T	133 E
167*R	243 C	279 C	308 E	317A T	362 C	391 T	396*R	414 C	420 C

LIQUID

39 T	40 E	42 E	45 E	69 E	70 E	82 E	86 E	98*R	99 R
108 T	127 T	133 E	138 R	146 R	147 T	150 T	154 T	167*R	182 E
183 C	235 T	244 C	255 C	279 C	286 E	287 E	308 E	322 E	328 C
362 C	366A R	385 E	395 T	396*R	420 C				

5. MELTING PRESSURE

49 E	50 E	75 E	92 T	145 T	229 E	270 E	313 E	319 E	337 E
347 E	348 E								

PROPERTY INDEX (CONT.)

6. SATURATION DENSITY

13 E 35 E 49 E 56 C 83 E 86 E 163 E 229 E 236 E 243 C
 256 E 258 E 265 E 317B T 383 E

SOLID (SOLID-LIQUID TRANSITION)

49 E 229 E

LIQUID (SOLID-LIQUID TRANSITION)

49 E 229 E 317B T

LIQUID (LIQUID-VAPOR TRANSITION)

13 E 35 E 56 C 83 E 86 E 236 E 243 C 256 E 258 E 265 E
 383 F

GAS (LIQUID-VAPOR TRANSITION)

83 E 86 E 163 E 236 E 243 C 256 E 258 E 265 E

7. LATENT HEATS

28 T 32 T 49 E 50 E 62 T 72 E 85 E 86 E 98*R 119 E
 122 E 125 R 126 T 132 E 133 E 137 E 147 T 152 T 155 T 183 C
 189*R 199 T 203 T 204A C 215A T 226 T 229 E 231 T 235 T 236 E
 243 C 248 T 255 C 257 E 279 C 300 C 314 T 315 T 316 T 318 C
 323*R 328 C 330 T 355 T 363*R 391 T 392 T 420 C

FUSION

49 E 50 E 62 T 72 E 98*R 122 E 125 R 132 E 133 E 147 T
 183 C 189*R 204A C 215A T 229 E 243 C 255 C 315 T 323*R 363*R

SUBLIMATION

28 T 32 T 85 E 98*R 189*R 199 T 314 T 316 T 391 T 420 C

VAPORIZATION

86 E 98*R 119 E 125 R 126 T 132 E 133 E 137 E 147 T 152 T
 155 T 183 C 189*R 203 T 204A C 226 T 231 T 235 T 236 E 243 C
 248 T 255 C 257 E 279 C 300 C 318 C 323*R 328 C 330 T 355 T
 363*R 392 T 420 C

PROPERTY INDEX (CONT.)

8. FIXED POINTS

1 T	62 T	70 E	72 E	73 E	74 T	79 C	82 E	83 E	91 T
92 T	98*R	125 R	133 E	137 E	138 R	138A E	146 R	147 T	153 C
159 T	162 R	182 E	183 C	196 C	204 C	215A T	234 T	235 T	236 E
240 T	243 C	255 C	256 E	256A R	265 E	270 E	273 E	279 C	286 E
287 E	295 C	298 R	301 E	302 T	303 T	308 E	318 C	328 C	344 T
358 T	362 C	363*R	395 T	396*R					

MELTING POINT (NORMAL)

62 T	72 E	73 E	98*R	125 R	159 T	183 C	243 C	255 C	279 C
286 E	303 T	318 C	363*R	395 T	396*R				

BOILING POINT (NORMAL)

70 E	74 T	98*R	125 R	137 E	138 R	138A E	146 R	183 C	235 T
243 C	255 C	279 C	286 E	287 E	302 T	318 C	328 C	363*R	396*R

TRIPLE POINT

70 E	72 E	73 E	91 T	92 T	98*R	133 E	137 E	138 R	138A E
146 R	182 E	204 C	256A R	270 E	273 E	279 C	301 E	363*R	396*R

CRITICAL POINT

1 T	79 C	82 E	83 E	91 T	98*R	125 R	146 R	147 T	153 C
162 R	196 C	207 C	215A T	234 T	236 E	240 T	243 C	255 C	256 E
265 E	279 C	286 E	287 E	295 C	298 R	308 E	318 C	328 C	344 T
358 T	362 C	363*R	395 T						

PROPERTY INDEX (CONT.)

10. VELOCITY OF SOUND

10 T	14 E	15 E	33 T	100 E	104 E	141 E	149 E	151 C	167*R
218 T	219 T	228 E	235A E	236 C	241 E	243 C	266 C	281 E	283 T
352A E	360 E	361 E	376 R	377 E	378 E	379 E	380 E	381 E	383 E
385 E	386 E	390 E	413 T						

SOLID

15 E	33 T	100 E	149 E	413 T
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LIQUID

10 T	104 E	141 E	151 C	241 E	283 T	376 R	377 E	378 E	379 E
380 E	383 E	385 E	386 E	390 E	413 T				

GAS (UP TO 200 DEGREES K)

236 E	266 C	380 E	381 E	413 T
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GAS (ABOVE 200 DEGREES K)

14 E	228 E	235A E	236 E	243 C	266 C	281 E	352A E	360 E	361 L
380 E	413 T								

11. JOULE-THOMSON COEFFICIENTS, INVERSION CURVE

22 T	98*R	135B C	172 T	222 T	223 T	236 E	266 C	269 C	321 E
385 E	401 C								

LIQUID

385 E

GAS (UP TO 200 DEGREES K)

135B C	236 E	266 C	321 E
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GAS (ABOVE 200 DEGREES K)

135B C	172 T	236 E	266 C	269 C	321 E	401 C
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PROPERTY INDEX (CONT.)

12. ENTROPY, ENTHALPY, INTERNAL ENERGY, FUGACITY,
GIBB'S FUNCTION, HELMHOLTZ FUNCTION

22 T	28 T	74 T	78 T	89 C	98*R	112 T	143 T	148 T	159 T
167*R	173 E	174 T	185 C	189*R	204A C	224 T	236 E	237 T	238 R
239 T	244 C	247 C	255 C	266 C	267 C	268 C	269 C	316 T	317 T
323*R	363*R	399 T	401 C	413 T	425 T				

SOLID

28 T	112 T	148 T	316 T	317 T	413 T
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LIQUID

174 T	204A C	238 R	317 T	399 T	413 T
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GAS (UP TO 200 DEGREES K)

78 T	143 T	174 T	224 T	236 E	237 T	238 R	266 C	399 T	413 T
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GAS (ABOVE 200 DEGREES K)

89 C	143 T	173 E	174 T	185 C	224 T	236 E	237 T	255 C	266 C
267 C	268 C	269 C	399 T	401 C	413 T				

PROPERTY INDEX (CONT.)

13. THERMAL CONDUCTIVITY

3 T	5 T	9 C	52 T	57 E	59 E	65 C	76 T	90 C	97 E
105 R	118 E	121 E	136 R	157 R	167*R	168 C	169 T	184 T	186 E
189*R	200 E	201 E	202 E	212 C	213 E	214 E	215 T	230 E	242 R
243 C	244 C	245 T	249A C	254 T	255 C	261 R	261A R	263 E	271 E
284 C	288 E	295 C	296 C	331 R	332 R	333 E	340 R	341 R	342 E
358 T	359 T	372 E	374 E	393 T	396*R	397 C	398 R	406 E	407 E
408 E	411 E	416 E	418 R	419 E	424 T				

SOLID

105 R	215 T	230 E	406 E	407 E	408 E				
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LIQUID

5 T	57 E	184 T	186 E	212 C	214 E	242 R	331 R	374 E	419 E
424 T									

GAS (UP TO 200 DEGREES K)

3 T	52 T	57 E	65 C	76 T	118 E	121 E	136 R	157 R	168 C
169 T	186 E	200 E	212 C	214 E	242 R	244 C	245 T	249A C	254 T
284 C	288 E	295 C	296 C	359 T	374 E	393 T	418 R	419 E	

GAS (ABOVE 200 DEGREES K)

3 T	9 C	52 T	59 E	65 C	90 C	97 E	118 E	121 E	157 R
168 C	169 T	186 E	200 E	201 E	202 E	212 C	213 E	214 E	242 R
243 C	244 C	245 T	249A C	254 T	255 C	261 R	263 E	271 E	284 C
288 E	295 C	296 C	332 R	333 E	340 R	341 R	342 E	359 T	372 E
393 T	398 R	411 E	416 E						

PROPERTY INDEX (CONT.)

14. VISCOSITY

3 T	5 T	6 T	8 T	9 C	17 R	36 T	43 T	44 E	52 T
53 T	65 C	77 T	79 C	93 T	114 T	118 E	121 E	124 C	125 R
126 T	129 E	129A E	134 E	135A E	157 R	167*R	168 C	169 T	170 T
187 E	189*R	191 E	195 T	196 C	206 E	207 E	208 E	209 E	210 C
211 C	212 C	225 E	240 T	242 R	244 C	249 E	251 R	254 T	255 C
259 T	261 R	264 E	271 E	279 C	282 T	288 E	297 R	309 E	310 E
320 E	325 C	326 C	327 E	335 E	336 E	339 E	345 C	358 T	359 T
366 C	366B T	367 R	382 E	387 E	387A E	388 E	389 E	396*R	400 C
403 T	417 E								

LIQUID

5 T	6 T	36 T	44 E	124 C	125 R	126 T	135A E	196 C	211 C
212 C	242 R	251 R	259 T	279 C	282 T	320 E	326 C	327 E	366 C
387 E	417 E								

GAS (UP TO 200 DEGREES K)

3 T	52 T	53 T	65 C	77 T	114 T	118 E	121 E	129 E	129A E
134 E	157 R	168 C	169 T	191 E	195 T	210 C	211 C	212 C	225 E
240 T	242 R	254 T	279 C	288 E	325 C	336 E	345 C	359 T	366 C
382 E	388 E	400 C							

GAS (ABOVE 200 DEGREES K)

3 T	8 T	9 C	17 R	43 T	52 T	53 T	65 C	77 T	79 C
114 T	118 E	121 E	129 E	129A E	134 E	157 R	168 C	169 T	170 T
187 F	191 E	195 T	206 E	207 E	208 E	209 E	211 C	212 C	225 E
240 T	242 R	244 C	249 E	254 T	255 C	261 R	264 E	271 E	279 C
288 E	297 R	309 E	310 E	325 C	335 E	336 E	339 E	345 C	359 T
366 C	367 R	387A E	388 E	389 E	400 C				

PROPERTY INDEX (CONT.)

15. PRANDTL NUMBER

167*R 168 C 212 C 271 E

LIQUID

212 C

GAS (UP TO 200 DEGREES K)

168 C 212 C

GAS (ABOVE 200 DEGREES K)

168 C 212 C 271 E

16. SELF-DIFFUSION COEFFICIENT

3 T	7 C	9 C	37 E	52 T	66 E	170 T	254 T	259 T	358 T
409 E	410 C								

SOLID

37 E

LIQUID

66 E 259 T

GAS (UP TO 200 DEGREES K)

3 T 52 T 254 T 409 E

GAS (ABOVE 200 DEGREES K)

3 T 7 C 9 C 52 T 254 T 409 E 410 C

PROPERTY INDEX (CONT.)

17. SURFACE TENSION

10 T	13 E	46 T	61 T	84 E	126 T	139 T	147 T	155 T	156 T
156A T	189*R	203 T	217 T	231 T	305 T	318 C	344 T	354 E	405 T

18. DIELECTRIC CONSTANT, CLAUSIUS-MOSSOTTI FUNCTION

47 E	51 C	54 E	71 E	92A T	116 E	117 E	158 E	188 E	189*R
190 E	252 C	258A T	260 E	261A R	272 E	275 C	294 E	421 E	

LIQUID

252 C	260 E
-------	-------

GAS (ABOVE 200 DEGREES K)

47 E	54 E	71 E	92A T	116 E	117 E	158 E	188 E	190 E	258A T
272 E	421 E								

19. REFRACTIVE INDEX

29 E	58 E	96 E	116 E	117 E	140 E	192 E	261A R	262 E	275 C
306 E	350 E	352 E	365A E	365B T					

SOLID

192 E	352 E
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LIQUID

192 E

GAS (ABOVE 200 DEGREES K)

29 E	58 E	96 E	116 E	117 E	140 E	262 E	365A E	365B T
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PROPERTY INDEX (CONT.)

20. LATTICE CHARACTERISTICS
(CRYSTAL STRUCTURE, LIQUID STRUCTURE, DEBYE TEMPERATURE)

28 T	32 T	33 T	91 T	102 C	105 R	112 T	148 T	161 E	165 T
181 C	199 T	277 C	349 E						

21. GRUNEISEN'S PARAMETER

101 E	106 T
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22. CORRESPONDING STATES

30 C	33 T	38 T	48 C	65 C	67 T	76 T	77 T	78 T	79 C
87 C	88 T	105 R	109 C	110 C	111 T	111A T	112 T	135B C	147 T
151 C	153 C	154 T	171 T	222 T	236 E	237 T	240 T	245 T	246 T
253 C	270 E	277 C	280 C	289 C	291 T	295 C	300 C	312 T	317A T
317B T	318 C	345 C	366B T	375 C	392 T	423 T			

23. INTERMOLECULAR POTENTIAL

12 T	89A T	102 C	170 T	254 T	302 T	303 T	314 T	316 T	353 T
404 T									

24. DOCUMENTS NOT APPEARING IN THE CROSS-INDEX

63 E	64 E	94 T	113 E	276 E	370 E	371 T
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COMPILATION - DATA FROM 177
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VAPORIZATION (133 TO 150 DEGREES K), ENTROPY, ENTHALPY, INTERNAL
ENERGY, SPECIFIC HEAT (P=CONSTANT)(V=CONSTANT), VELOCITY OF SOUND,
JOULE-THOMSON COEFFICIENT (GAS) (133 TO 273 DEGREES K AND 1 TO
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COMPRESSIBILITIES OF GASES.
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300. PITZER, K. S.
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- TRIPLE POINT TEMPERATURE AND PRESSURE
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- BOILING POINT, INTERMOLECULAR FORCES
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303. POPOVICI, S. AND POP, M.
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- SPECIFIC HEAT (V=CONSTANT)(LIQUID, GAS)(90 TO 190 DEGREES K)
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 ANN. PHYSIK VOL. 74, 255-74 (1924)
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- DENSITY (GAS) (285 TO 292 DEGREES K AND 535 TO 767 MM HG),
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- VISCOSITY (GAS)(285 DEGREES K)
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310. RAYLEIGH
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- DENSITY, VISCOSITY (GAS)(273 DEGREES K)
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C.A. 40, 4271 4
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- 317B. RIEDEL, L.
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 KORRESPONDENZPRINZIPI.
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 OF 0.4 TO 1.0), THEORY OF CORRESPONDING STATES, MELTING AND BOILING
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 AN EXPERIMENTAL DETERMINATION OF THE MELTING CURVES OF ARGON AND
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320. ROBINSON, D. W.
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 C.A. 53, 15680 F
- VISCOSITY (LIQUID) (90 DEGREES K AND 1 TO 2000 ATM)
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321. ROEBUCK, J. R. AND OSTERBERG, H.
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- JOULE-THOMSON COEFFICIENT (GAS) (123 TO 573 DEGREES K AND 1 TO 200 ATM)
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327. RUDENKO, N. S. AND SCHUBNIKOV, L. W.
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(VISCOSITY OF LIQUID NITROGEN, CARBON MONOXIDE, ARGON AND OXYGEN AND ITS DEPENDENCE ON TEMPERATURE.)
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C.A. 29, 2040 4
- VISCOSITY (LIQUID)(84 TO 87 DEGREES K)
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ANN. PHYSIK VOL. 29, 751-79 (1909)
- P-V-T DATA (GAS)(284.2 AND 510.3 DEGREES K), VAPOR PRESSURE (85 DEGREES K), BOILING POINT, CRITICAL POINT, HEAT OF VAPORIZATION, EQUATION OF STATE
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ZHUR. EKSPITL. I TEORET. FIZ. VOL. 24, 107-13 (1953) (IN RUSSIAN)
C.A. 49, 3589 A
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330. RYKOV, V. I.
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C.A. 55, 12019 B
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- THERMAL CONDUCTIVITY (LIQUID)(88 TO 198 DEGREES K)
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- THERMAL CONDUCTIVITY (GAS)(276.5 DEGREES K AND 0.1 TO 593.3 MM HG)
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333. SCHAFFER, K.
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 C.A. 54, 1002 H
- THERMAL CONDUCTIVITY (GAS)(273 TO 1373 DEGREES K)
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334. SCHAMES, L.
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 C.A. 25, 1130 7
- EQUATION OF STATE, VIRIAL COEFFICIENTS (GAS)(173 TO 673 DEGREES K)
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 VEREINIGTEN FRIEDRICHS UNIVERSITY, HALLE, PH. D. THESIS (1909)
- VISCOSITY (GAS)(288 DEGREES K AND 1 ATM)
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336. SCHMITT, K.
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ANN. PHYSIK VOL. 30, 393-410 (1909)
- VISCOSITY (GAS)(90 TO 456 DEGREES K AND 73 TO 76 CM HG)
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DIE WARME VOL. 54, 97-8 (1931)
C.A. 25, 1708 2
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338. SCHULTZE, H.
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ANN. PHYSIK. VOL. 48, 269-72 (1915)
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340. SCHWARZE, W.
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PHYSIK. Z. VOL. 3, 264 (1902)
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342. SCHWARZE, W.
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- ACCOMMODATION COEFFICIENT, MOLECULAR HEAT (GAS)(287 DEGREES K)
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344. SHIMOKAWA, J.
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- CRITICAL TEMPERATURE, PARTITION FUNCTION, CELL-SURFACE TENSION,
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 VISCOSITY-REDUCED STATE CORRELATION FOR THE INERT GASES.
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- VISCOSITY (45 TO 15000 DEGREES K), PRINCIPLE OF CORRESPONDING STATES
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346. SIMON, F. AND KIPPERT, F.
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Z. PHYSIK. CHEM. VOL. 135, 113-28 (1928)
- ISOMETRIC SLOPE (SOLID, LIQUID)(72 TO 90 DEGREES K), DENSITY (SOLID)
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347. SIMON, F. E., RUHEMANN, M. AND EDWARDS, W. A. M.
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Z. PHYSIK. CHEM. (LEIPZIG) VOL. B6, 331-42 (1930)
C.A. 24, 3411 2
- MELTING CURVE(SOLID,LIQUID)(83 TO 150 DEGREES K AND 0 TO 3380
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348. SIMON, F. E., RUHEMANN, M. AND EDWARDS, W. A. M.
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C. A. 24, 4681
- MELTING CURVE
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349. SIMON, F. AND VON SIMSON, C.
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Z. PHYSIK VOL. 25, 160-4 (1924)
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REFRACTIVE INDEX OF SOLID KRYPTON AND SOLID ARGON.
PHIL. MAG. VOL. 6, 939-42 (JUL 1961)
C.A. 56, 9569 H
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J. CHEM. PHYS. VOL. 38, 825-7 (FEB 1963)
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